1 Stable isotope constraints on Holocene carbon cycle changes from an 2 Antarctic ice core

Joachim Elsig¹, Jochen Schmitt^{1,2}, Daiana Leuenberger¹, Robert Schneider¹, Marc
Eyer¹, Markus Leuenberger¹, Fortunat Joos¹, Hubertus Fischer^{1,2} & Thomas F.
Stocker¹

⁷
 ⁸ ¹Climate and Environmental Physics, Physics Institute, University of Bern,
 ⁹ Sidlerstrasse 5, CH-3012 Bern, Switzerland, and Oeschger Centre for Climate
 ¹⁰ Change Research, University of Bern, Switzerland.

- ²Alfred Wegener Institute for Polar and Marine Research (AWI), Columbusstrasse, D 27568 Bremerhaven, Germany
- 13

14 Reconstructions of atmospheric CO_2 based on Antarctic ice cores^{1, 2} reveal significant concentration changes during the Holocene, but the processes 15 16 responsible for these CO₂ changes have not been unambiguously identified. Distinct characteristics in the carbon isotope signatures of the major carbon 17 reservoirs (ocean, biosphere, sediments, and atmosphere) constrain variations 18 19 in the CO₂ fluxes between those reservoirs. Here, we present the first highly resolved atmospheric δ^{13} C record for the last 11 kyr measured on atmospheric 20 CO₂ trapped in an Antarctic ice core. Based on mass-balance inverse model 21 22 calculations^{3, 4} performed with a simplified carbon cycle model, we show that 23 the decrease in atmospheric CO₂ of about 5 ppmv and the increase in δ^{13} C of about 0.25‰ during the early Holocene is most probably the result of a 24 combination of carbon uptake of about 290 GtC by the land biosphere and 25 carbon release from the ocean in response to carbonate compensation of the 26 27 terrestrial uptake during the termination of the last ice age. The 20 ppmv increase of atmospheric CO₂ and the small decrease in δ^{13} C of about 0.05‰ 28 29 during the later Holocene are dominated by contributions from carbonate 30 compensation of earlier land biosphere uptake and coral reef formation, but 31 only by a small decrease of the land biosphere carbon inventory.

32

33 The Holocene is the current interglacial period and started about 11 kyr BP (11,000 34 years before present, where present is defined as AD 1950) following the Transition 35 (here defined as 18-11 kyr BP) from the last glacial maximum (LGM). Variations in the atmospheric concentration of CO₂ during the Holocene were significant but small 36 37 compared to glacial-interglacial changes of typically 100 ppmv (parts per million by volume)^{5, 6}. Yet, a decrease of about 5 ppmv from 11-7.5 kyr BP could be observed 38 followed by an increase of about 20 ppmv to the pre-industrial level of about 280 39 ppmv^{1, 2, 7}. Different explanations for these variations were discussed^{7, 8} such as 40 changes in the carbon inventories of vegetation, soils and peatlands⁹, in anthropogenic land use^{10, 11}, in sea surface temperature^{7, 12}, coral reef growth^{13, 14} or 41 42 carbonate compensation¹⁵. The latter is a multi-millennial equilibration process of the 43 atmosphere-ocean-sediment system and the weathering cycle. Moreover, model 44 simulations of atmospheric CO₂ and δ^{13} C during the Holocene did not provide an 45 unambiguous quantitative explanation^{7, 8, 16}. The major stumbling block has been the 46 scarcity of reconstructions of δ^{13} C on atmospheric CO₂ with sufficient accuracy and 47 time resolution from the LGM to the Holocene^{7, 17-20}. 48

49

50 We performed carbon isotope measurements on air trapped in the EPICA (European 51 Project for Ice Coring in Antarctica) Dome C (75°06′S, 123°24′E) ice core using two

1 completely independent extraction methods (mechanical cracking and sublimation. see supplementary information). Altogether, 199 single samples have been 2 measured from 59 different depths in the interval of 110 to 410 meters. This interval 3 corresponds to a gas age range of 11-0.35 kyr BP²¹. The results are gravitationally 4 corrected (see Supplementary Information) and are presented in Figure 1. Of the 165 5 6 samples which were extracted with a mechanical cracker, a minimum of two were 7 taken at each depth. For the remaining 34 samples we used a sublimation technique, and they are either single measurements or replications of three adjacent ice 8 9 samples. The overall precision for a single measurement is 0.07‰ for both methods. 10 The results of the two different methods agree very well within their uncertainties. The record clearly shows a continuous increase in the δ^{13} C values during the first 5 11 kyr of the Holocene followed by only slightly decreasing values. 12

13

14 We focus on the evolution of the carbon isotopes on a time scale of a few thousand 15 years. Therefore, a spline and its 1σ uncertainty bands have been calculated with a cut-off period of 5 kyr (Figure 2). In a Monte Carlo simulation, standard deviations 16 17 smaller than 0.07‰ were increased to 0.07‰; those higher than 0.07‰, however, were retained. Measurements at one depth interval (open symbols at 2.5 kyr BP in 18 Figure 1) led to exceptionally negative δ^{13} C values significantly outside the 2σ 19 uncertainty range independently for both extraction methods. The reason for these 20 21 outliers remains obscure but the very large scatter in neighbouring samples despite the low-pass filtering effect of the bubble-enclosure process rules out an atmospheric 22 23 origin of these outliers (see Supplementary Information). Accordingly, these two outliers are not included in the calculation of the spline. 24

25

26 Two main features of the carbon isotope record can be recognized: first, an almost 27 linear increase from -6.58‰ to -6.33‰ between 11-6 kyr BP, and second, a small 28 decrease of 0.05‰ in the later Holocene after 6 kyr BP. In Figure 2 our data are compared with two published δ^{13} C records from the Taylor Dome⁷ and Law Dome²² 29 ice cores. The relative timing of the Taylor Dome to Dome C gas age was adapted 30 31 from Monnin et al.¹, who matched the CO₂ records of Dome C and Taylor Dome for the Holocene. The Taylor Dome data generally agree with our measurements except 32 33 for 3 data points (out of 12) that have no overlap within the reported 1σ uncertainties 34 of our record. Our data are entirely consistent with the Law Dome record covering the 35 last 1,000 years providing a direct link to direct atmospheric measurements through the firn gas data from Law Dome²². 36

37

We use our new, more precise and better resolved δ^{13} C record to quantify carbon 38 fluxes with different isotopic signatures and to test various hypotheses of Holocene 39 CO_2 variations. Previous mechanistic model studies^{8, 14, 16, 23}, ocean sediment analyses¹⁵, coral reef reconstructions²⁴, and peatland data^{9, 25} suggest that the 40 41 42 Holocene CO₂ variations were caused by (i) land biosphere uptake, (ii) carbonate compensation of earlier land biosphere uptake, and/or (iii) the build-up of coral reefs 43 made of calcium carbonate. In contrast, changes in sea surface temperature (SST)¹², 44 45 ocean circulation and marine biological cycling, shifts in the proportion of C_3 to C_4 photosynthesis⁸, or changes in volcanic outgassing are assumed to be of minor 46 importance for the Holocene CO₂ and δ^{13} C evolution. In addition, a strong release of 47 carbon over the last 7,000 years from anthropogenic land use changes has been 48 previously proposed¹⁰. The multitude and the spatio-temporal variability of the 49 processes influencing atmospheric CO₂ and δ^{13} C prevent a firm attribution of the 50 measured changes to a single mechanism. However, the deconvolution of our new 51

1 record by mass balance inverse calculations^{3, 4, 7} (see Supplementary Information) 2 permits us to quantify atmosphere-ocean and atmosphere-land carbon fluxes for 3 different hypotheses and to attribute the measured CO_2 and $\delta^{13}C$ changes to the 4 most likely mechanisms.

5

6 Model calculations were performed with a cost-efficient impulse response 7 representation of carbonate compensation and of the High-Latitude 8 ocean Exchange/Interior Diffusion-Advection model coupled 4-box to а 9 representation of vegetation and soils and a well mixed atmosphere. The substitute 10 model yields results for atmospheric CO₂ and the redistribution of carbon and carbon isotopes between global reservoirs that are comparable to those of spatially resolved 11 12 models. For instance for the main scenario of land carbon uptake discussed below, 13 the substitute yields a late Holocene CO₂ rise of 15 ppmv compared to 12 ppmv 14 obtained with the Bern3D dynamic ocean-sediment model.

15

16 First, a land-biosphere only scenario is assessed. Atmospheric CO₂ variations are assumed to be entirely driven by changes in the land biosphere (using a fractionation 17 for photosynthesis of 18.7³, and the ocean-sediment system to react only 18 19 passively. Solving the atmospheric CO₂ budget for the unknown terrestrial flux yields a land biosphere uptake of 75 GtC from 11-7 kyr BP and a release of 275 GtC 20 thereafter⁷. More importantly, the simulated evolution of atmospheric δ^{13} C is not 21 compatible with the δ^{13} C measurements (Figure 2). The release of isotopically 22 depleted terrestrial carbon yields a modelled δ^{13} C decrease of 0.25‰ after 8 kyr BP, 23 whereas our data show little change. Thus, suggestions that CO₂ emissions from 24 anthropogenic land use change caused the late Holocene CO₂ rise¹⁰ are 25 quantitatively inconsistent with our δ^{13} C record, as well as with other evidence⁸. 26

27

Other scenarios considering only one driving mechanism (changes in SST, the 28 29 marine biological cycle, or the calcium carbonate cycle only), or the combination of marine biological changes and land biosphere changes, are also in conflict with the 30 measured co-evolution of CO₂ and δ^{13} C and with other proxy evidence. For example, 31 an unrealistically large global average SST increase of 2.5°C is required to explain 32 33 the δ^{13} C increase of 0.25‰ from 11-6 kyr BP by SST changes only. A combination of SST and land biosphere changes (tested by deconvolving both atmospheric records 34 simultaneously) implies a global average SST increase of about 1.5°C, which is in 35 conflict with alkenone-based SST reconstructions (0.2 \pm 0.2°C since 8 kyr BP²⁶) and 36 model calculations^{12, 27}. This is different from earlier findings by Indermühle et al.⁷ 37 38 who applied the same modelling approach but had to rely on only very few δ^{13} C 39 values. They suggested that changes in the land biosphere with a modest 40 temperature increase, possibly in combination with changes in the marine calcium 41 carbonate cycle are responsible for the Holocene CO_2 evolution. Our improved $\delta^{13}C$ 42 record permits us now to exclude this land biosphere-global SST scenario.

43

Finally, we turn to the land biosphere-marine carbonate scenario which assumes that atmospheric variations are caused by changes in the land biosphere in combination with carbonate compensation and coral reef growth. Both marine carbonate processes are associated with small isotopic fractionations and cannot be distinguished by δ^{13} C data nor can they significantly influence atmospheric δ^{13} C. Solving the two atmospheric budgets for CO₂ and δ^{13} C yields a land biosphere uptake of (290 ± 36) GtC (mean and ±1 σ confidence interval from a Monte Carlo analysis) during 11-5 kyr BP and a release of (36 ± 37) GtC thereafter (Figure 3a). Note that the substitute model represents global mean fluxes and is not able to capture the influence of spatial variations in carbon and isotopic exchange. Potential contributions from other processes such as volcanism, SST changes or changes in the marine biological cycle are neglected in the land biosphere-marine carbonate scenario.

7

8 Are these fluxes inferred from our mass balance calculation realistic and in 9 agreement with proxy data, model results, and process understanding? The early 10 Holocene land biosphere uptake of 290 GtC is compatible with an early Holocene uptake of 110 GtC simulated by the Lund-Potsdam-Jena Dynamic Vegetation model 11 on non-peatland land in response to reconstructed ice-sheet retreat, climate, and 12 CO₂ variations⁸. This requires an early Holocene increase in peat carbon by 180 GtC. 13 14 Current inventories of peat carbon range from 270 to 455 GtC⁹. Radiocarbon dating 15 suggests initiation of peat accumulation already during the Transition with a major expansion of peat area in the early Holocene⁹, but also some peat accumulation in 16 recent millennia²⁸. A land biosphere release of \sim 40 GtC in the later Holocene may be 17 18 explained by (i) a model-estimated release of 30 GtC due to the desertification of the Sahara⁷, (ii) a modelled release of around 50 GtC from pre-1,500 anthropogenic land 19 use change¹¹, and (iii) an implied peat carbon uptake of ~40 GtC. The implied 20 Holocene peat carbon uptake of 220 GtC is somewhat lower than the published 21 range for the current carbon inventory in peat carbon⁹. This might be explained by 22 peat carbon uptake during the Transition. Alternatively, natural carbon release from 23 land might have been larger than 30 GtC²³ implying also a higher uptake by peat. 24

25 In a next step the atmosphere-ocean component of our carbon cycle model was used 26 in forward simulations to test mechanistic explanations of inferred ocean-driven CO₂ 27 changes. First, we prescribe the atmosphere-land fluxes from the land biospherecarbonate compensation scenario to separate the contribution from Holocene land 28 biosphere changes. The atmospheric $\delta^{13}C$ record is matched by design. The 29 Holocene land biosphere changes result in an initial decrease in atmospheric CO₂ of 30 31 about 17 ppmv, much larger than measured in ice cores, and an increase of ~9 ppmv 32 after 6.5 kyr BP (Figure 3b). The increase is attributed to carbonate compensation of 33 the early Holocene land biosphere uptake (~6 ppmv) and to the (prescribed) land 34 biosphere release of 50 GtC (~3 ppmv). Second, we consider that carbonate compensation caused by the land biosphere uptake during the Transition also 35 36 contributed to the Holocene CO₂ rise. Land biosphere carbon inventory changes over 37 the glacial-interglacial period have been estimated to be around 500 to more than 1,000 GtC²⁹, ³⁰. Here, the land biosphere inventory is prescribed to increase linearly 38 39 by 700 GtC during the Transition. This yields a simulated early Holocene CO₂ decrease comparable to the ice core data and a late Holocene rise of ~15 ppmv. 40 41 somewhat smaller than reconstructed (see Supplementary Information for additional 42 sensitivity simulations).

43

The remaining ~5 ppmv in atmospheric CO_2 can be plausibly explained by coral reef growth. Coral reef formation changes the carbonate ion balance in sea water and increases atmospheric CO_2^{13} . Vecsei and Berger²⁴ reconstructed a calcium carbonate accumulation by reefs of 260 GtC during the past 8 kyr. This implies a CO_2 rise by another ~8 ppmv¹⁴, compatible with the ice core data and model results. However, our modeling results and the $\delta^{13}C$ record do not quantitatively support the

- 1 suggestion that coral reef growth caused a much larger CO₂ rise by 20 to 40 ppmv¹⁴
- 2 during the late Holocene.
- 3

High-resolution and high-precision ice core data on atmospheric δ^{13} C are able to 4 5 significantly constrain the possible pathways of the carbon cycle evolution in the past. Based on carbon cycle modeling our new $\delta^{13}C$ data show that the evolution of 6 7 atmospheric CO₂ in the past 11 kyr is dominated by an early Holocene increase in 8 land biosphere and changes in the marine calcium carbonate cycle. Based on our 9 δ^{13} C record suggestions that CO₂ emissions from anthropogenic land use changes caused the later Holocene CO_2 rise¹⁰ and prevented a new ice age are not tenable. 10 The attribution of the CO₂ changes to specific changes in the marine carbonate cycle 11 12 (including coral reef growth, carbonate compensation of land biosphere carbon 13 uptake, sediment-ocean interactions related to the reorganization of the marine 14 carbon cycle during the glacial-interglacial transition, or changes in weathering 15 fluxes) is challenging. Further progress requires an extension of our atmospheric 16 δ^{13} C record into the glacial period as well as process studies with 3-dimensional physical-biogeochemical climate models. 17

1 References

- Monnin, E. et al. Atmospheric CO₂ concentrations over the last glacial termination. Science **291**, 112-114 (2001).
- 5 2. Flückiger, J. et al. *High-resolution Holocene* N_2O *ice core record and its* 6 *relationship with* CH_4 *and* CO_2 . Global Biogeochemical Cycles **16**, 10.1-10.8 7 (2002).
- 3. Joos, F. & Bruno, M. Long-term variability of the terrestrial and oceanic carbon
 sinks and the budgets of the carbon isotopes ¹³C and ¹⁴C. Global
 Biogeochemical Cycles **12**, 277-295 (1998).
- Bruno, M. & Joos, F. Terrestrial carbon storage during the past 200 years: A
 Monte Carlo analysis of CO₂ data from ice core and atmospheric
 measurements. Global Biogeochemical Cycles **11**, 111-124 (1997).
- Lüthi, D. et al. *High-resolution carbon dioxide concentration record 650,000-*800,000 years before present. Nature **453**, 379-382 (2008).
- 16 6. Sigman, D. M. & Boyle, E. A. *Glacial/interglacial variations in atmospheric* 17 *carbon dioxide*. Nature **407**, 859-869 (2000).
- 187.Indermühle, A. et al. Holocene carbon-cycle dynamics based on CO2 trapped19in ice at Taylor Dome, Antarctica. Nature **398**, 121-126 (1999).
- Joos, F., Gerber, S., Prentice, I. C., Otto-Bliesner, B. L. & Valdes, P. J.
 Transient simulations of Holocene atmospheric carbon dioxide and terrestrial carbon since the Last Glacial Maximum. Global Biogeochemical Cycles 18, 1 18 (2004).
- 9. MacDonald, G. M. et al. Rapid early development of circumarctic peatlands and atmospheric CH_4 and CO_2 variations. Science **314**, 285-288 (2006).
- Ruddiman, W. F. *The anthropogenic greenhouse era began thousands of years ago*. Climatic Change **61**, 261-293 (2003).
- Strassmann, K. M., Joos, F. & Fischer, G. Simulating effects of land use
 changes on carbon fluxes: past contributions to atmospheric CO₂ increases
 and future commitments due to losses of terrestrial sink capacity. Tellus
 Series B-Chemical and Physical Meteorology **60**, 583-603 (2008).
- Brovkin, V., Kim, J. H., Hofmann, M. & Schneider, R. A lowering effect of
 reconstructed Holocene changes in sea surface temperatures on the atmospheric CO₂ concentration. Global Biogeochemical Cycles 22, GB1016
 (2008).
- Berger, W. H. Increase of Carbon-Dioxide in the Atmosphere during
 Deglaciation the Coral-Reef Hypothesis. Naturwissenschaften 69, 87-88
 (1982).
- Ridgwell, A. J., Watson, A. J., Maslin, M. A. & Kaplan, J. O. Implications of
 coral reef buildup for the controls on atmospheric CO₂ since the Last Glacial Maximum. Paleoceanography 18, 1083-1092 (2003).
- 42 15. Broecker, W. S., Lynch-Stieglitz, J., Clark, E., Hajdas, I. & Bonani, G. What
 43 *caused the atmosphere's CO₂ content to rise during the last 8000 years?*44 Geochemistry Geophysics Geosystems 2, art. no.-2001GC000177 (2001).
- 45 16. Brovkin, V. et al. Carbon cycle, vegetation, and climate dynamics in the
 46 Holocene: Experiments with the CLIMBER-2 model. Global Biogeochem.
 47 Cycles 16(4), (86-1)-(86-13) (2002).
- 48 17. Smith, H. J., Fischer, H., Wahlen, M., Mastroianni, D. & Deck, B. *Dual modes*49 of the carbon cycle since the Last Glacial Maximum. Nature 400, 248-250
 50 (1999).

- 18. Leuenberger, M., Siegenthaler, U. & Langway, C. C. Carbon Isotope
 Composition of Atmospheric CO₂ during the Last Ice-Age from an Antarctic Ice
 Core. Nature **357**, 488-490 (1992).
- Eyer, M. Highly resolved δ¹³C measurements on CO₂ in air from Antarctic ice cores. PhD Thesis, Climate and Environmental Physics, Physics Institute, University of Bern, 1-113 (2004).
 Lourantou, A. Contraindre l'augmentation en dioxyde de carbone (CO₂) lors
- 720.Lourantou, A. Contraindre l'augmentation en dioxyde de carbone (CO2) lors8des déglaciations basés sur son rapport isotopique stable du carbone9 $(\delta^{13}CO_2)$. PhD Thèse de l'Université Joseph Fourier, Grenoble 1, France10(2008).
- 11 21. Loulergue, L. et al. *New constraints on the gas age-ice age difference along* 12 *the EPICA ice cores, 0-50 kyr.* Climate of the Past **3**, 527-540 (2007).
- 1322.Francey, R. J. et al. A 1000-year high precision record of δ^{13} C in atmospheric14CO2. Tellus Series B-Chemical and Physical Meteorology **51**, 170-193 (1999).
- Schurgers, G. et al. Dynamics of the terrestrial biosphere, climate and
 atmospheric CO₂ concentration during interglacials: a comparison between Eemian and Holocene. Climate of the Past **2**, 205-220 (2006).
- Vecsei, A. & Berger, W. H. Increase of atmospheric CO₂ during deglaciation:
 Constraints on the coral reef hypothesis from patterns of deposition. Global
 Biogeochemical Cycles 18, GB1035 (1-7) (2004).
- 21 25. Smith, L. C. et al. Siberian peatlands a net carbon sink and global methane
 22 source since the early Holocene. Science **303**, 353-356 (2004).
- 23 26. Kim, J. H. et al. North Pacific and North Atlantic sea-surface temperature
 24 variability during the holocene. Quaternary Science Reviews 23, 2141-2154
 25 (2004).
- 26 27. Wang, Y., Mysak, L. A. & Roulet, N. T. Holocene climate and carbon cycle
 27 dynamics: Experiments with the "green" McGill Paleoclimate Model. Global
 28 Biogeochemical Cycles 19, GB3022 (2005).
- 29 28. Beilman, D. W., MacDonald, G. M., Smith, L. C. & Reimer, P. J. Carbon
 30 accumulation in peatlands of West Siberia over the last 2000 years. Global
 31 Biogeochem. Cycles 23, GB1012 (2009).
- 32 29. Duplessy, J. C. et al. *Deepwater source variations during the last climatic*33 *cycle and their impact on the global deepwater circulation*. Paleoceanography
 34 3, 343–360 (1988).
- 35 30. Crowley, T. J. *Ice-Age Terrestrial Carbon Changes Revisited*. Global
 Biogeochemical Cycles 9, 377-389 (1995).
- 37
- 38 39

1 **Supplementary Information** is linked to the online version of the paper at

2 <u>www.nature.com/nature</u>.

3

4 Acknowledgements This work is a contribution to the European Project for Ice 5 Coring in Antarctica (EPICA), a joint European Science Foundation/European 6 Commission scientific programme, funded by the EU (EPICA-MIS) and by national 7 contributions from Belgium, Denmark, France, Germany, Italy, the Netherlands, Norway, Sweden, Switzerland and the United Kingdom. The main logistic support 8 9 was provided by IPEV and PNRA (at Dome C) and AWI (at Dronning Maud Land). 10 We thank A. Landais, D. Rodriguez, E. Capron and G. Dreyfus for the contribution of δ^{15} N data as well as P. Nyfeler and K. Grossenbacher for their technical support, T. 11 12 Tschumi for sharing his Bern3D results, and J. Chappellaz for comments. We acknowledge financial support by the Swiss NSF, the DFG priority program 13 INTERDYNAMIK and the German climate program DEKLIM. This is EPICA 14 15 publication no. 227.

16

Author contribution J.E., J.S., D.L., R.S., and M.E. performed the measurements.
 F.J. performed modelling and interpretation. M.L., H.F., and T.F.S designed
 research. All authors participated in discussions on method development,
 interpretation and presentation of results.

21

Author information Reprints and permissions information is available at <u>www.nature.com/reprints</u>. Correspondence and requests for materials should be addressed to T.F.S. (stocker@climate.unibe.ch).

- 25 26

1 Figure legends

2

3 Figure 1

 δ^{13} C and CO₂^{1, 2} measured in air trapped in ice from Dome C, Antarctica. Blue triangles indicate measurements performed with the cracker (mean of two to four samples), red circles measurements with the sublimation method (single measurements or mean of three adjacent samples). Open symbols indicate outliers. The error bars represent the t-weighted 1*σ* standard deviations of the mean (s.d.m). Grey squares represent CO₂ data from Dome C (mean of six samples; error bars, 1*σ* of the mean)²¹.

11

12 **Figure 2**

 δ^{13} C ice core records measured on the Antarctic ice cores from Dome C. Taylor 13 **Dome⁷ and Law Dome²².** Grey, yellow, and green colored symbols refer to results 14 15 from Dome C, Taylor Dome and Law Dome, respectively. Grey open symbols 16 indicate outliers. The error bars for Dome C measurements are the same as in Fig. 1. The error bars for the Taylor Dome and Law Dome data represent 1σ of the mean 17 18 (s.d.). The grey line through the Dome C data represents the mean of one hundred 19 Monte Carlo simulations with a cut-off period of 5 kyr. The grey shaded area 20 indicates the 1σ standard deviation of the spline (s.d.). The red line is the result of a 21 deconvolution of the atmospheric CO₂ record assuming the land biosphere scenario.

22

23 Figure 3

Attribution of simulated CO₂ to different processes. (a) Cumulative land 24 biosphere uptake for the land biosphere-marine carbonate scenario from 25 deconvolving the atmospheric $CO_2^{1, 2}$ and $\delta^{13}C$ records. The shaded area indicates 26 the 1σ confidence interval of a Monte Carlo analysis⁴, taking into account the 27 28 uncertainty of the ice-core data. (b) Atmospheric CO₂ is simulated by prescribing the 29 land biosphere changes shown in (a). Grey squares represent CO₂ data from Dome 30 C (mean of six samples; error bars, 1σ of the mean). Green line: land biosphere only; black line: land biosphere and carbonate compensation (carb. comp.) during the 31 32 Holocene; red line: including 700 GtC land biosphere uptake during the Transition. 33 The bar chart indicates the modelled contributions to the CO₂ rise from 6.5-0 kyr BP by individual processes assuming the land biosphere-marine carbonate scenario. 34 35 The remaining difference between the simulated and measured CO₂ increase may be 36 attributed to coral reef growth and other mechanisms.

1 Methods summary

For the cracker method we use a mechanic extraction device (steel needle cracker), 2 3 which crushes the ice sample (6 g) under vacuum to release the enclosed gases in 4 the ice. In case of the sublimation method the gases are extracted from the ice 5 sample (30 g) using sublimation at -25°C in a glass vessel. In either case the 6 released air is dried in a water trap and the amount of air is measured with a 7 pressure gauge to calculate the CO₂ concentration. CO₂ and N₂O are separated from the major air components at -196°C using liquid nitrogen with subsequent 8 9 concentration by means of a cryofocus capillary to ensure complete gas chromatographic separation. After having passed a chromatographic column to 10 separate isobaric components (N₂O and organic components from drilling fluid) the 11 12 purified CO₂ sample is injected via an open-split into the isotope ratio mass spectrometer (a Delta Plus XL for the cracker method and a MAT 253 for the 13 14 sublimation method, both ThermoFisher). Both extraction systems are equipped with 15 inlet devices which allow processing calibrated reference gases in the same way as ice samples. This allows checking the system and referencing the results from the ice 16

17 samples on an international standard.



Stable isotope constraints on Holocene carbon cycle changes from an Antarctic ice core

Elsig et al.



Stable isotope constraints on Holocene carbon cycle changes from an Antarctic ice core

Elsig et al.

Figure 3



Stable isotope constraints on Holocene carbon cycle changes from an Antarctic ice core

Elsig et al.

Supplementary Information

to

Stable isotope constraints on Holocene carbon cycle changes from an Antarctic ice core

Joachim Elsig, Jochen Schmitt, Daiana Leuenberger, Robert Schneider, Marc Eyer, Markus Leuenberger, Fortunat Joos, Hubertus Fischer & Thomas F. Stocker

13 14 15

12

5 6 7

8

9 10 11

16 **Cracker method:**

One sample requires 5-6 g of ice (V = $l \cdot w \cdot h$ = 2.2.1.5.2.2 cm³). This corresponds to 17 about 0.5 ml STP of air or 0.1 µl STP CO₂. A stainless steel needle cracker is lined 18 19 up with the preconcentration system (Precon) and a Delta Plus XL mass 20 spectrometer (MS) from ThermoFisher allowing online measurements. The ice cracker consists of a stainless steel body (57 ml) in which an array of stainless steel 21 22 needles is mounted to crack the ice under vacuum at a temperature of -20°C. The 23 released gas expands through a water trap (-70°C) into a small volume, where the 24 gas pressure to evaluate the CO₂ concentration is measured. The extracted air is 25 then flushed by a high helium flux of about 600 ml/min through the cracker to the Precon. This unit permits the quantitative separation of CO₂ and N₂O from air as well 26 27 as switching from a high to a low helium stream (~1 ml/min). After having passed a 28 GC (gas chromatography) column to separate CO₂ from N₂O and possible organic 29 components, e.g. derived from drilling fluid, the CO₂ sample is injected into the MS via an open-split. During one day of measurements several runs with bubble free ice 30 (EK) combined with a gas of known isotopic composition (EG II, $\delta^{13}C = -4.74\%$ 31 32 VPDB) are performed. These standard gas measurements allow checking the 33 system and referencing the samples¹.

34 35

36 Sublimation method:

The retrieval of δ^{13} C values on CO₂ from gas enclosures in ice cores using sublimation (together with δ^{18} O and mixing ratios of CO₂ and N₂O) is split into two separate analytical systems: First, the gas extraction using sublimation in a vacuum line and secondly, a sample clean-up in a helium flow line coupled to continuous flow isotope ratio mass spectrometry (CF-IRMS).

The principle behind the gas extraction is sublimation allowing for a quantitative 42 43 release of air trapped in either bubble or clathrate ice samples. Within a glass vessel, 44 a cylindrical ice sample of 30 g is held at -25°C via a cold air stream. By illuminating 45 the sample with infrared light a water vapour flux is established from the sample to a nearby cold trap thereby guantitatively releasing the enclosed gases. This gas 46 47 stream is dried in a cold trap at -120°C and afterwards CO₂ and N₂O are separated from the major air components (N_2 , O_2 , Ar) at -196°C. The amount of the major air 48 49 components is measured with a pressure gauge connected to a temperature 50 controlled expansion volume. Together with the signal from the mass spectrometer the mixing ratios of CO_2 and N_2O can be calculated by referencing it to the standard gas concentrations. CO_2 and N_2O are transferred into a glass tube, which is then flame-sealed to be stored until the clean-up and measurement step. The extraction line is equipped with a gas inlet to continuously inject calibrated air samples into the sublimation vessel, thus, mimicking the continuous gas release during the sublimation of an ice sample. These air samples are used as a reference for both the isotopic measurement and the mixing ratio of CO_2 and to monitor the entire analysis.

The sample clean-up line consists of a tube cracker device to open the sealed glass 8 tubes within a helium flux of 0.9 ml min⁻¹, a cryofocus capillary to generate a sharp 9 sample peak and a chromatographic column to separate isobaric components (N₂O 10 and organic components from drilling fluid) from CO_2 . Finally, pure CO_2 is then 11 12 admitted to a Finnigan 253 IRMS (isotope ratio mass spectrometer) via an open split interface. This off-line set-up minimizes additional uncertainty introduced by the clean 13 up and IRMS measurement procedure as all samples can be measured at the 14 15 identical instrument conditions thus, day to day instrument variations are omitted. Like the gas extraction line, the sample clean-up line is equipped with a reference 16 device to introduce working standards of CO₂/N₂O prior to each ice or air sample. 17 18 The peak size of the working standard can be adjusted to cover the range of sample sizes thereby accounting for a potential dependency of the δ^{13} C signal on CO₂ 19 20 amount².

- 21
- 22
- 23

24 Discussion of the outlier at 2.5 kyr BP

The samples with a gas age of 2,519 yr BP were not taken into account for the 25 interpretation and for calculating splines. Air from this depth was extracted 26 independently with the cracker as well with the sublimation method. Both methods 27 indicate a much too negative δ^{13} C value. A closer look on the raw data of this depth 28 interval revealed that replicate ice samples only a few cm apart (or a few years in 29 age) show δ^{13} C variations of 0.4‰. Given the broad age distribution of the enclosed 30 air (around 170 years) we can clearly rule out an atmospheric perturbation as cause. 31 As the CO₂ concentration measured on the same samples deviate only by a few 32 33 ppmv from neighbouring depth intervals a post drilling contamination with recent atmospheric air (with more negative δ^{13} C values and elevated CO₂ concentration) 34 can be ruled out as well. However, just 0.3 m above this interval drilling problems 35 occurred (E. Wolff, personal communications) likely generating extreme mechanic 36 37 stress on the ice below which can be responsible for these outliers, yet the 38 underlying process is not understood.

39

40

41

42 **Gravitation correction**:

43 As mentioned in the text, the isotopes of the enclosed air are fractionated due to 44 gravitational settling. As a result, the heavy isotopes are concentrated at the bottom of the firn column. Accordingly, the enclosed air in the ice is enriched in heavy 45 46 molecules. This fractionation depends on the mass difference of the gas species, the 47 mean site temperature and on the diffusive column height of the firn. Since the isotopic composition of atmospheric N₂ remained constant over long time intervals, 48 measuring δ^{15} N in air from ice cores can be used to correct δ^{13} C for the gravitation 49 effect. Thereby, changes in δ^{15} N can directly be applied to correct δ^{13} C values since 50

the mass differences of the involved isotopes are the same. Existing $\delta^{15}N$ data for 1 Dome C are shown in Figure S1 (b). The single δ^{15} N value at 0 yrs BP is taken from 2 firn measurements performed within the framework of the EU project FIRETRACC³. 3 4 δ^{15} N values from 1 to 9 kyr BP are unpublished values measured at LSCE in 2008 (D. Rodriguez, E. Capron and A. Landais, personal communication), and $\delta^{15} N$ values 5 after 9 kyr BP were measured at LSCE⁴ in 2003. Also shown in Figure S1 are two 6 splines of our data set, one through the uncorrected data and the other through the 7 8 gravitation corrected data. The influence of the gravitation correction is summarised 9 in Table ST1:

10 The main difference is the time when the δ^{13} C maximum is reached (T_{max}). Since the 11 gravitation correction is almost constant during the Holocene, the main conclusions 12 are not affected by this correction. Therefore, the increase in δ^{13} C by about 0.25‰ in 13 the early Holocene and the slight decrease in the late Holocene are robust.

- 14
- 15
- 16

17 Comparison of extraction methods (cracker and sublimation):

Since the measurements were performed with two different extraction methods and 18 19 two distinct mass spectrometers and separate standard gases, the two data sets 20 have to be checked for a possible offset. Therefore, splines were calculated for both records (Figure S2). Additionally, Table ST2 summarises the main features of the two 21 records including a comparison between the slopes from 11-8 kyr BP and 6-0 kyr BP. 22 23 The main differences between the two records are the point of time, T_{max} , where the maximum δ^{13} C value is reached as well as the strength of the decrease after this 24 point. Both records suggest an increase of δ^{13} C between 11-8 kyr BP. Within the 25 error of the slopes, this increase is robust. The decrease after 6 kyr BP is slightly 26 27 higher for the sublimation measurements but it is also consistent within the 28 measurements errors.

- 29
- 30

31

32 Comparison with a former δ^{13} C record derived with the cracker method

Another unpublished δ^{13} C record over the Holocene period was already established in 2004 with a cracker system similar to the one used in this study but with a significantly higher scatter⁵. A compilation of all three records is shown in Figure S3.

Within the range of error, the data derived earlier are in good agreement with the 36 37 data presented in the main text except for the period from about 8 to 7 kyr. Here, the 38 older measurements are systematically too low by about 0.4‰. So far we have not been able to find any explanation for this deviation. Since this early study, however, 39 40 substantial progress has been made by constructing an improved and dedicated 41 cracker system, and by including a gas chromatographic column which helped to 42 significantly reduce the measuring uncertainty. Therefore, we refrain from including this data set into the discussion in the main text. 43

44

45 46

47 Splines for different cut-off periods:

In order to quantify the effect of different cut-off periods, splines⁶ were calculated for cut-off periods between 0.5 kyr and 6 kyr (Figure S4). For periods between 0.5-1 kyr,

50 the splines are not smooth but show millennial oscillations. Considering our

1 uncertainty of $\pm 0.07\%$, and the time resolution of our data, these oscillations are not 2 significant and cannot be interpreted. However, it is noteworthy that the Law Dome 3 ice core data⁷ show a wiggle that would be in line with millennial oscillations 4 suggested by our new Holocene record. Data with further improved precision and 5 time resolution are needed to verify or falsify such millennial-scale oscillations. For a 6 cut-off period of more than 3 kyr, the oscillations disappear and the long-term trend 7 as mentioned in the text becomes visible.

8 9

10

11 Mass-spectrometric CO₂ measurements:

With both of our measuring set-ups it is possible to estimate the CO₂ concentration 12 parallel to δ^{13} C values. The CO₂ concentration can be calculated by comparing the 13 ratio of the gas extraction pressure and the peak area of mass to charge ratio 44 in 14 the mass spectrometer with the corresponding ratio of a standard gas. The mean 15 16 reproducibility for the cracker and sublimation methods is 2.6 ppmv and 2.9 ppmv, respectively. This is about twice as much as for the dedicated CO₂ concentration 17 method⁸. Concerning the cracker measurements the main uncertainty lies in the 18 19 behaviour of the water vapour and desorption/adsorption effects during the cracking 20 process and the subsequent pressure measurement. Water vapour may account for a small pressure increase and possibly affects the pressure readings. In case of the 21 22 concentration results from the sublimation method the pressure reading is sensitive 23 to temperature fluctuations which may explain the larger scatter and slightly higher 24 values. The results are presented in Figure S5 in comparison with the measurements 25 performed on the EDC96 ice by a laser absorption spectroscopy method⁹.

26 27

2829 Model description and additional results

30

31 Model description

32 Model calculations were performed with an impulse response representation of the High-Latitude Exchange/Interior Diffusion-Advection (HILDA) ocean model coupled to 33 a 4-box representation of vegetation and soils and a well-mixed atmosphere¹⁰. 34 35 Carbonate compensation is included by assuming that a fraction, a, of 70% of the terrestrial release (uptake) is absorbed by sediments on an exponential timescale, τ , 36 of 5 kyr^{11, 12}. A ¹³C/¹²C fractionation of 18.7‰ is applied for the atmosphere-to-land 37 38 biosphere flux and fractionation for fluxes between atmosphere and surface ocean are from Mook¹³ as described elsewhere¹⁴. 39

40

41 *Mass-balance inverse model calculations*

Carbon fluxes in the land biosphere-only scenario (S₁) are quantified by a (single) 42 deconvolution of the atmospheric CO₂ record¹⁵. Atmospheric CO₂ is prescribed and 43 44 carbon uptake by the ocean-sediment system is simulated. The total carbon inventory in the atmosphere-land biosphere-ocean-sediment system remains 45 constant. Then, the net land biosphere-to-atmosphere flux is equal to the prescribed 46 47 change in the atmospheric carbon inventory and the calculated ocean uptake flux. The land biosphere-global SST (SST: sea surface temperature) scenario (S_2) , the 48 land biosphere-marine biosphere scenario (S₃), and the land biosphere-marine 49 50 carbonate compensation scenario (S₄) are quantified by (double) deconvolutions of

page 5

the atmospheric CO₂ and δ^{13} C records as detailed by Joos and Bruno¹⁴ and 1 2 Indermühle et al.¹¹. In the land biosphere-global SST scenario, the partial pressure of 3 CO₂ in the surface ocean is forced to be consistent with the inferred net sea-to-air flux (from the mass balance of CO_2 and ${}^{13}CO_2$) and the prescribed atmospheric CO_2 4 5 by adjusting global SST in the model. The isotopic fractionation for the air-sea and 6 sea-air fluxes is evaluated for the adjusted global SST. In the land biosphere-marine 7 biosphere scenario, the surface ocean CO₂ partial pressure is similarly adjusted by adding (removing) dissolved inorganic carbon that is depleted in δ^{13} C by 20% relative 8 to the surface ocean ¹³C to ¹²C ratio. In the land biosphere-carbonate compensation 9 scenario, the surface partial pressure is adjusted without modification of the ¹³C to 10 12 C ratio. The small fractionation of ~1‰ that occurs during calcium carbonate 11 12 formation is neglected.

13 The land biosphere-global SST scenario S₂ yields a net carbon flux of about 190 GtC to the terrestrial biosphere from 11-6 kyr BP and a decrease of the terrestrial 14 15 biosphere of 120 GtC from 6-0 kyr BP (Figure S6). Derived global SST indicates an 16 increase by about 1.5°C over the Holocene, which is unrealistic. The double deconvolution for the land biosphere-marine biosphere scenario S₃ leads to 17 18 unrealistically high terrestrial carbon fluxes and requires an increase of the terrestrial 19 biosphere of about 700 GtC from 11-3.5 kyr BP followed by a decrease of 100 GtC 20 from 3.5-0 kyr BP. Such large fluxes and the large increase in the terrestrial 21 biosphere during the Holocene are most unlikely and in contrast to vegetation model estimates¹² and pollen-based reconstructions of vegetation cover¹⁶. However, we can 22 not entirely exclude that marine biological processes have contributed to the co-23 evolution of atmospheric CO₂ and δ^{13} C. 24

25

26 Sensitivity of simulated CO₂ to the timescales of carbonate compensation and the 27 temporal evolution terrestrial carbon uptake during the transition

Model simulations were performed to project atmospheric CO₂ from prescribed land 28 29 biosphere-atmosphere fluxes to evaluate whether the results of the mass balance 30 inverse calculation for the land biosphere-carbonate scenario are plausible. The 31 sensitivity of simulated CO₂ to changes in the following parameters is addressed: 32 carbonate compensation timescale τ ; the fraction *a* of the perturbation to be mitigated 33 by carbonate compensation, and the uptake history of the terrestrial biosphere. Four 34 different schematic uptake histories prior to the Holocene are selected and shown in 35 Figure S7: (i) no land biosphere uptake during the Transition to detect the effect of the carbonate compensation caused by the land biosphere uptake during the early 36 37 Holocene; (ii), the standard uptake that assumes a linear increase in the land biosphere inventory of 700 GtC during the Transition in 7 kyr (constant flux of 0.1 38 39 GtC yr⁻¹ from 18 to 11 ka BP); (iii), a rapid land biosphere uptake of 700 GtC in 2 kyr; and (iv), a slow land biosphere uptake of 700 GtC in 10 kyr. The CO₂ decrease in the 40 early Holocene is stronger and the increase during the late Holocene is smaller for 41 42 the 10 kyr uptake history compared to the standard. A rapid but intense uptake 43 history during the Transition (iii) results in a stronger increase of CO₂ during the late Holocene. The measured atmospheric CO₂ data are simulated better when τ is set to 44 7 kyr and the fractionation a is increased to 0.8. Good agreement with the data in 45 these still very simplified scenarios results for $\tau = 8$ kyr, a = 0.8 and assuming a rapid 46 and large land biosphere uptake during the period 13 to 11 kyr BP. This reflects a 47 scenario with a small role of coral reef buildup for Holocene CO₂. 48 49

1 References:

2 3 Elsig, J. New insights into the global carbon cycle from measurements of CO₂ 1. 4 stable isotopes: methodological improvements and interpretation of a new 5 EPICA Dome C ice core δ^{13} C record. PhD Thesis, Climate and Environmental Physics, Physics Institute, University of Bern, 1-174 (2009). 6 7 Schmitt, J. A sublimation technique for high-precision δ^{13} C on CO₂ and CO₂ 2. 8 mixing ratio from air trappes in deep ice cores. PhD Thesis, Fachbereich 9 Geowissenschaften, Universität Bremen, 1-167 (2006), 10 3. FIRETRACC. in Part C: Scientific Report (European Commission DGX, Environment and Climate, 2000). 11 12 4. Dreyfus, G. Dating an 800,000 year antarctic ice core record using the isotopic 13 composition of trapped air. PhD thesis, Departement of Geoscience, Princeton University (2008). 14 Eyer, M. Highly resolved δ^{13} C measurements on CO₂ in air from Antarctic ice 15 5. cores. PhD Thesis, Climate and Environmental Physics, Physics Institute, 16 17 University of Bern, 1-113 (2004). Enting, I. G. On the Use of Smoothing Splines to Filter CO₂ Data. Journal of 18 6. 19 Geophysical Research-Atmospheres 92, 10977-10984 (1987). Francey, R. J. et al. A 1000-year high precision record of δ^{13} C in atmospheric 20 7. 21 CO₂. Tellus Series B-Chemical and Physical Meteorology **51**, 170-193 (1999). 22 8. Lüthi, D. et al. High-resolution carbon dioxide concentration record 650,000-23 800,000 years before present. Nature 453, 379-382 (2008). 24 9. Monnin, E. et al. Atmospheric CO₂ concentrations over the last glacial 25 termination. Science 291, 112-114 (2001). Joos, F. et al. An efficient and accurate representation of complex oceanic and 26 10. 27 biospheric models of anthropogenic carbon uptake. Tellus Series B-Chemical and Physical Meteorology 48, 397-417 (1996). 28 Indermühle, A. et al. Holocene carbon-cycle dynamics based on CO₂ trapped 29 11. in ice at Taylor Dome, Antarctica. Nature 398, 121-126 (1999). 30 31 12. Joos, F., Gerber, S., Prentice, I. C., Otto-Bliesner, B. L. & Valdes, P. J. 32 Transient simulations of Holocene atmospheric carbon dioxide and terrestrial 33 carbon since the Last Glacial Maximum. Global Biogeochemical Cycles 18, 1-34 18 (2004). Mook, W. G. ¹³C in Atmospheric CO₂. Netherlands Journal of Sea Research 35 13. 36 **20**, 211-223 (1986). 37 14. Joos, F. & Bruno, M. Long-term variability of the terrestrial and oceanic carbon 38 sinks and the budgets of the carbon isotopes 13 C and 14 C. Global 39 Biogeochemical Cycles 12, 277-295 (1998). 40 15. Bruno, M. & Joos, F. Terrestrial carbon storage during the past 200 years: A Monte Carlo analysis of CO₂ data from ice core and atmospheric 41 42 measurements. Global Biogeochemical Cycles 11, 111-124 (1997). 43 16. Prentice, I. C., Jolly, D. & participants, B. Mid-Holocene and glacial-maximum 44 vegetation geography of the northern continents and Africa. J. Biogeogr 27, 45 507-519 (2000). 46 47 48

Elsig et al.

1 Tables:

2 3 4

	δ ¹³ C _{min} (‰)	δ ¹³ C _{max} (‰)	T _{max} (‰)	δ ¹³ C _{end} (‰)	(11- <i>T_{max}</i>) kyr BP increase (‰)	(<i>T_{max}-T_{end}</i>) kyr BP decrease (‰)
(1) uncorrected	-6.07	-5.83	6116	-5.89	0.24	-0.06
(2) corrected	-6.58	-6.33	5535	-6.38	0.25	-0.05

5

ST1: Summary of the main features of the spline through the uncorrected data set (1) 6 and through the gravitation corrected data set (2). T_{max} is the year BP where the spline through the data reaches its maximum δ^{13} C value whereas $\delta^{13}C_{end}$ refers to 7

8 9 the value of the spline for the most recent sample.

10

11

12

	δ ¹³ C _{min} (‰)	δ ¹³ C _{max} (‰)	T _{max} (‰)	δ ¹³ C _{end} (‰)	11-8 kyr BP increase (‰/kyr)	6-0 kyr BP decrease (‰/kyr)
(1) cracker	-6.60	-6.32	5750	-6.35	-0.06±0.02	0.01±0.01
(2) sublimation	-6.64	-6.32	4800	-6.43	-0.07±0.03	0.018±0.009

13

ST2: Summary of the main features of the spline through the cracker data record (1) 14 and through the sublimation data record (2) as well as the comparison between the 15 16 linear slopes through the individual records and their uncertainties for the intervals 11-8 kyr BP and 6-0 kyr BP. T_{max} is the year BP where the spline reaches its δ^{13} C 17 maximum whereas $\delta^{13}C_{end}$ refers to the value of the spline for the most recent 18 19 sample.





4

Figure S1: The effect of the gravitation correction on the δ^{13} C record. (a) The blue 5 line is the spline through the uncorrected data set (left v-axis) with its 1σ uncertainty 6 7 band (dashed line), whereas the red line corresponds to the spline through the gravitation corrected data set (right y-axis) with its corresponding 1σ band. (b) The 8 inset corresponds to $\delta^{15}N$ data. The single $\delta^{15}N$ value at 0 yrs BP is taken from firm 9 measurements (black cross)³. δ^{15} N values from 1 to 9 kyr BP are unpublished values 10 measured at LSCE by D. Rodriguez, E. Capron and A. Landais in 2008 (black 11 diamonds), and δ^{15} N values after 9 kyr BP were measured at LSCE by G. Drevfus in 12 2003 (black squares). A spline through the $\delta^{15}N$ data was used to correct the $\delta^{13}C$ 13 data. This spline indicates a mean correction of (0.50 ± 0.01) %. Within the error 14 band, the two splines through the δ^{13} C data agree. Therefore, the main conclusions 15 of this paper are not affected by the gravitation correction. 16 17



1

Figure S2: Spline through the cracker data set (blue line) with 1σ error band and the spline through the sublimation data set (red line). The grey shaded area is the 1σ band of the spline through both data sets. The two splines agree within their uncertainties. Therefore, an offset of the methods can be excluded.



1

Figure S3: Comparison of the δ^{13} C data from this study with unpublished data (green diamonds, mean of mostly two samples, error bars are 1σ of the mean) established in 2004 in Bern⁵ with a preliminary cracker method compared to the one used in this study.



1

Figure S4: Splines of Monte-Carlo simulations with different cut-off periods. For cutoff periods longer than about 3 kyr, the oscillating behaviour of the splines disappears and the long-term trend becomes robust. The dashed red lines indicate the 1σ band of the spline with a cut off period of 5 kyr.



1

Figure S5: CO_2 record of air trapped in Dome C ice. Blue triangles are the measurements performed with the cracker set-up. Red circles are CO_2 concentration measurements obtained using the sublimation set-up. The error bars represent the tweighted 1σ standard deviation of the mean. Grey squares are the CO_2 measurements from Monnin et al.⁹. The black line is the spline through the record of Monnin et al. with a cut-off period of 5 kyr and the grey shaded area is the corresponding 1σ band.



Figure S6: Cumulative land biosphere uptake from inverse modelling results for the
 discussed scenarios: (S₁: land-biosphere-only scenario, S₂: land biosphere-global
 SST scenario, S₃: land biosphere-marine biota scenario, S₄: land biosphere-marine
 carbonate compensation scenario).



1

3 Figure S7: Sensitivity tests for the CO₂ evolution predicted by the land biosphere-4 marine carbonate compensation scenario S₄ during the Holocene. a) During the 5 Transition, the cumulative land biosphere uptake of 700 GtC is prescribed by a linear 6 increase in 7 kyr (thick black line, (ii)), in 2 kyr (dashed black line, (iii)) or in 10 kyr (dash-dotted black line, (iv)). After 11 kyr BP, the land biosphere uptake is dictated 7 8 by the output of S₄. The black thin line shows the case where no terrestrial uptake 9 took place over the Transition (i). b) The line character is given by the different 10 uptake histories presented left whereas the colour describes the influence of 11 changing τ and a.